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The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N_2 are mixed with 12.0 mol of H_2 according to the following equation: $N_2(g) + 3H_2(g) \dots$

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Chapter 9 Stoichiometry Worksheet Mole review Worksheet. Calculate the molar mass of the following, and then write the molar mass in ratio form. KNO_3 .

Na_2SO_4 . $\text{Ca}(\text{OH})_2$ $(\text{NH}_4)_2\text{SO}_3$. CuSO_4 .

Moles and Mass. Determine the number of moles in each of the quantities below.

Use the factor label method. ...

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